

# 1 **Safe use of Electric Arc Furnace Dust (EAFD) as secondary raw** 2 **material in self-compacting mortars production**

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## 19 **Abstract**

20 This research contributes to the development of the Circular Economy concept by managing  
21 waste through its use as a construction material. A novel process in which two samples of

22 industrial waste Electric Arc Furnace Dust (EAFD) collected from two different steelwork  
23 plants are used as secondary raw material (SRM) in the production of self-compacting mor-  
24 tars is proposed. At a previous stage, a characterisation of mortar material components was  
25 carried out. Then, self-compacting mortars with replacement ratios of 25%, 50% and 100%  
26 of siliceous filler with EAFD were designed. The feasibility of EAFD use was analysed by  
27 means of the study of mechanical strength, mineralogical, capillary properties and leaching  
28 behaviour in monolithic and granular state. The environmental classification of mortars, ac-  
29 cording to their leaching behaviour, was carried out in accordance with European Directive  
30 2003/33/EC (2003). EAFD incorporation modified the self-compactability and common hy-  
31 dration reactions. Mechanical strength decreased with EAFD incorporation, although the  
32 mortar with 25% of replacement with one of the EAFD's showed a negligible loss compared  
33 to the self-compacting reference mortar. Water absorption by capillarity increased with  
34 EAFD content, consistently with the mortars' mechanical behaviour. The leaching behaviour  
35 analysis in a monolithic state revealed that all mortar releases were below the permitted limit.  
36 In a granular state, the mortar with the best mechanical behaviour was environmentally safe,  
37 leaving all the elements encapsulated by the self-compacting matrix. The results of this study  
38 contribute to EAFD management through its valorisation as secondary raw material in the  
39 production of self-compacting mortars, addressing an unprecedented line of research.

40 ***Keywords:***

41 Electric Arc Furnace Dust; self-compacting mortars; secondary raw material; mechanical  
42 behaviour mortar; leaching behaviour mortar; construction material.

43 **1. Introduction**

44 The Circular Economy concept describes a cyclical system in which economic and environ-  
45 mental aspects are integrated, with the objective of reducing both raw materials consumption

46 and generated waste. The Circular Economy Implementation Plan of the European Parlia-  
47 ment Commission (COM(2017)33, 2017) has highlighted the special importance of waste  
48 management and of the use of secondary raw materials (SRM). Therefore, the recovery of  
49 waste through its use as SRM is an attractive alternative in the fight against the scarcity of  
50 natural resources.

51 The construction sector, specifically the concrete industry, stands out for its special role  
52 within the Circular Economy, since this material is one of the most used and consumes  
53 many resources (Rattanashotinunt et al., 2018). At a European level, the construction sec-  
54 tor is considered one of the main sources of waste. Due to this, the use of waste in concrete  
55 manufacturing is a goal that has been pursued for decades. The wastes traditionally used  
56 are fly ash, electric arc furnace slag (Anastasiou et al., 2014; Kim et al., 2018) and con-  
57 struction and demolition waste (López-Uceda et al., 2016; Silva et al., 2016), generating  
58 beneficial situations in this industry.

59 Additionally, self-compacting concrete technology (SCC) has been one of the most im-  
60 portant advances in construction materials. These fluid mixes can be placed and com-  
61 pacted under their own weight, without vibration, segregation, blockage of coarse aggre-  
62 gate, bleeding, or paste exudation (EFNARC, 2005). Because their use has been wide-  
63 spread in recent years in civil engineering works, buildings and the precast industry, al-  
64 ternative materials rather than conventional ones have gained relevance in concrete pro-  
65 duction. In accordance with this, a wide range of wastes, such as filler of bituminous  
66 mixes (Esquinas et al., 2017), filler from marble and tile waste (Tennich et al., 2015) and  
67 electric arc furnace slag (Santamaría et al., 2017), has been investigated successfully, ob-  
68 taining more environmentally-friendly self-compacting concrete.

69 During steel production by scrap melting as a raw material, metals such as Zn, Pb, Fe or

70 Cd are volatilised to later react with oxygen condensing into solid particles (Issa et al.,  
71 2012). These particles are named Electric Arc Furnace Dust (EAFD). Due to its high  
72 content of potentially leachable heavy metals, several directives such as the European  
73 Waste Catalogue (EWC, 2002) and the United States Environmental Protection Agency  
74 (USEPA, 2018) classify it as hazardous waste. The production of EAFD is estimated at  
75 around 15-25 kg per ton of steel produced (Sayadi and Hesami, 2017). Between 5 and 7  
76 million tons of EAFD are produced worldwide (Kelebek et al., 2004).

77 Directly recycling EAFD in the furnace is difficult because the recirculation of volatile ele-  
78 ments increases production costs and produces an accumulation of impurities in the melted  
79 steel mass (Nyirenda, 1991). In addition, EAFD is made of extremely fine particles (0.1 - 200  
80  $\mu\text{m}$ ) (Lenz and Martins, 2007) and its handling is difficult without previous pelleting.

81 Globally, 70% of EAFD is landfilled, after treatment (Maslehuddin et al., 2011). Stabili-  
82 zation/Solidification (S/S) techniques are used in waste encapsulation, particularly for  
83 those containing heavy metals (Chen et al., 2009). Therefore, S/S is the technique gener-  
84 ally used in the management of EAFD. The effectiveness of this process is studied by  
85 physical, chemical and microstructural methods, analysing the efficiency degree mainly  
86 through two parameters: mechanical strengths and leaching behaviour of the solids gener-  
87 ated (Ledesma et al., 2017).

88 Although studies about S/S of EAFD have been successful (Laforest and Duchesne, 2006;  
89 Salihoglu and Pinarli, 2008), landfilling is not an option compatible with the current paradigm  
90 of Circular Economy, due to the use of natural resources during the treatment, such as natural  
91 aggregates or cement, and the content of valuable metals lost in this process.

92 In the search for alternative uses of EAFD to landfill, Sayadi and Hesami (2017) evalu-  
93 ated its effectiveness as a filler in asphalt mixes and Cholake et al. (2018) combined blue

94 slate aggregate and EAFD in the production of high-strength lightweight aggregate for  
95 use as construction material. Other authors, such as da Silva Magalhães et al. (2017),  
96 report that the utilisation of EAFD in the construction industry is an interesting option  
97 compared to landfilling, as it allows encapsulating hazardous elements in a stable matrix.

98 According to self-compactability properties, SCC is denser than conventional concrete,  
99 which is why, this matrix could allow the encapsulation of hazardous waste type EAFD.

100 The need to have a greater volume of ultrafine material in the SCC could be combined  
101 with the benefit of contributing to the elimination of the EAFD. This alternative has not  
102 been investigated until now.

103 The aggregate content in SCC is lower than that of conventional concrete; so, part of the  
104 performance of SCC is defined by the behaviour of the self-compacting mortar (SCM)  
105 phase, one of the main components of SCC (Khotbehsara et al., 2018). Previous evalua-  
106 tion of the mortar phase to analyse the characteristics of SCC has been carried out by  
107 different authors, such as Nepomuceno et al. (2012) or da Silva et al. (2015).

108 Our study analyses the influence of incorporating two EAFD wastes as filler replacement  
109 on the performance of SCM, evaluating the safety of their use through the mechanical  
110 and leaching behaviour. This research has as objective to develop economic and sustain-  
111 able techniques in construction and to search viable EAFD management, reducing the  
112 consumption of natural resources for its treatment and avoiding deposit in hazardous  
113 waste landfills. To the best of the authors' knowledge, this is the first research that pro-  
114 duced SCM mechanically and environmentally safe, achieving EAFD/CEM ratios higher  
115 than those available up to now. This line of research has not been previously explored  
116 and promotes the development of the new Circular Economy paradigm.

## 117 2. Experimental details

### 118 2.1. Material characterisation

119 The materials used in the manufacturing of SCM were: two EAFD wastes, Portland ce-  
120 ment, siliceous filler, two fractions of siliceous natural sand and a superplasticiser.

121 The wastes, named EAFD1 and EAFD2, came from two steelworks located in Olaberria  
122 and Zumarraga (Guipúzcoa, Spain). Both steelworks use an Electric Arc Furnace (EAF)  
123 for steel production. The EAFD samples were collected according to UNE-EN  
124 14899:2007 (AENOR, 2017) from the electrostatic precipitators of the Electric Arc Fur-  
125 naces. The cement used was CEMI/42.5R (CEM), supplied by SECIL Company (Lisbon,  
126 Portugal). The siliceous filler (SF) was provided by Minas Carmina Palau Saverdera (Ge-  
127 rona, Spain). Two fractions of aggregates were used in the mortars manufacture, fine sand  
128 0/2 and coarse sand 0/4 (NS-0/2 and NS-0/4). The MasterEase 5025 (S<sub>p</sub>) admixture was  
129 supplied by BASF Company. This is a superplasticiser/high-activity reducer of water for  
130 the production of high-strength, low viscosity and improved rheology concrete. The S<sub>p</sub>  
131 structure is based on PAE polymers (polyacrylic ester) and has some analogy with that of  
132 polycarboxylates, but has important differences in the composition of the main chain.

133 The chemical composition of EAFD and NS was carried out by X-ray fluorescence (XRF)  
134 using 4 kW of power and S4PIONEER, BRUKER equipment. The CEM and SF compo-  
135 sition was supplied by their manufacturers.

136 To obtain the X-ray diffraction (XRD) patterns of both EAFD, a device named Bruker  
137 D8 Discover A25 with Cu-K $\alpha$  radiation was used. The goniometric scan was studied from  
138 10° to 70° (2 $\theta$ ) at a speed of 0.00625° min<sup>-1</sup>.

139 Both EAFD were analysed by Fourier transform infrared (FTIR), by Bruker Tensor 27-  
140 Hyperion 2000 spectrophotometer (Bruker Optik GmbH, GERMANY) with CsI beam

141 splitters and a DTGS detector. The transmission spectrum was collected using OPUSv  
142 6.5 software.

143 The specific surface area of both EAFD and SF was analysed by the Brunauer-Emmett-  
144 Teller method (BET), using Micromeritics ASAP 2010 equipment. The EAFD specific  
145 gravity was determined by UNE 80103:2013 (AENOR, 2017). Regarding NS-0/2 and  
146 NS-0/4, the specific gravity and water absorption was calculated according to UNE-EN  
147 1097-6:2014 (AENOR, 2017). The specific surface area of CEM and specific gravity of  
148 CEM and SF were provided by the manufacturers.

149 The particle size distribution of EAFD, CEM and SF was measured using a MastersizerS  
150 device (Malvern Instruments) with ethanol as a dispersant of the samples. The particle size  
151 distribution of NS-0/2 and NS-0/4 was tested in accordance with UNE-EN 933-1:2012  
152 (AENOR, 2017). In addition, electron microscopy (SEM) was performed on both EAFD.  
153 This technique was carried out by a scanning electron microscope JEOL JSM-6300.

154 EAFD1 and EAFD2 were analysed by means of the compliance test according to UNE-  
155 EN 12457-4:2003 (AENOR, 2017) to study their leaching behaviour on the levels of  
156 heavy metal release, listed in the European Directive 2003/33/CE (2003).

157 After the compliance tests, the eluates were analysed by Mass Spectrometry with Induc-  
158 tively Coupled Plasma (ICP-MS) using a Perkin Elmer ELAN DRC-e spectrometer. The  
159 leaching results obtained for the compliance test were compared with the legal limits es-  
160 tablished by European Directive 2003/33/CE (2003) on the acceptance of landfill waste,  
161 attached in Table 1, which allows environmental classification of materials based on the  
162 release of their components.

## 163 **2.2. Mortar mix proportions**

164 A total of 7 mixes were manufactured in this study. All SCM were designed using the

165 Nepomuceno method (2005) according to the following criteria:

166 - A value of 0.70 was chosen for the volumetric ratio ( $V_p/V_s$ ), in absolute volume,  
167 in all mortars. This ratio was the total amount of powder materials ( $V_p$ ) and fine  
168 aggregates in the mix ( $V_s$ ). Nepomuceno et al. (2005) proposed volumetric ratio  
169 values between 0.60 and 0.80;

170 - With the aim of using EAFD as SF replacement, a reference mortar (SCRM) was  
171 designed with 50% of CEM and 50% of SF. The mortars with waste proportions  
172 were produced by replacing SF by EAFD in volume percentages ( $f_{Ad}$ ) of 25%,  
173 50% and 100%. Mortars with EAFD were named SCM-EAFDi-j:

174 ○  $i$  = EAFD type (EAFD1 or EAFD2);

175 ○  $j$  = the replacement percentage of SF by EAFD;

176 - In this study, self-compactability was controlled via the relative spread area ( $G_m$ )  
177 using the slump-flow test, and iteratively adjusting the  $V_w/V_p$  and  $S_p/p\%$  ratios.  
178 This methodology has been explained in several previous works (Nepomuceno,  
179 2005; Nepomuceno et al., 2012; Silva et al., 2011).  $V_w/V_p$  was the ratio in absolute  
180 volume between the water and the powder materials and  $S_p/p\%$  was the ratio in  
181 percentage between the amount in mass of superplasticiser ( $S_p$ ) and powder ma-  
182 terials ( $p$ ).  $G_m$  was calculated according to equation (1) proposed by Nepomuceno  
183 et al. (2012) and EFNARC (2005).

184 - 
$$G_m = \left(\frac{D_m}{D_0}\right)^2 - 1 \quad (1)$$

185 - Where  $D_m$  is the slump-flow average of the diameters recorded during the test and  
186  $D_0$  the initial diameter of the base of the cone used in the test.  $D_m$  was kept in the  
187 range of  $270 \pm 30$  mm according to the values proposed by Kanadasan et al.  
188 (2018), which was established as suitable for SCM.

189 Table 2 shows the composition of the mixes, the nomenclature used in this research and



190 the parameters applied to achieve self-compactability of the mortars. In addition, Table 3  
191 shows the amount of constituents used per m<sup>3</sup> of mortar.

192 All mortars were mixed by the same procedure using a standard mixer (AENOR, 2017).  
193 The constituents were homogenised in solid state for 30 seconds. Afterwards, 80% of  
194 water was added. This paste was mixed for 2.5 minutes. Then, the remaining 20% of  
195 water was added with the Sp diluted and mixing continued for 1 minute. All these steps  
196 occurred at low velocity (140 rpm). After that, it was stopped for 1 minute and finally it  
197 continued for 1 minute at a higher velocity (285 rpm).

198 From each mix, a total of 12 prismatic specimens (40 mm x 40 mm x 160 mm) were  
199 made. The specimens were stored in their mould in a climatic chamber with 95% ± 5%  
200 of relative humidity and 20 °C ± 2 °C of temperature for 24 hours, after which they were  
201 demoulded and cured under the same conditions until testing.

### 202 **2.3. Test methods**

203 The flexural and compressive strengths of the hardened mortar were studied in accord-  
204 ance with UNE-EN 1015-11:2000 (AENOR, 2017) at 7, 28 and 91 days using 3 speci-  
205 mens per age.

206 XRD was used to analyse crystalline mineral components. For that purpose, hardened  
207 mortars were crushed and sieved through a 0.125 mm sieve. The goniometric exploration  
208 employed swept from 5° to 80° (2θ°) at a speed of 0.0142° min<sup>-1</sup>.

209 The water absorption by capillarity was determined in mortars at atmospheric pressure  
210 according to UNE-EN 1015-18: 2003 (AENOR, 2017) at 28 days of age. For this test 6  
211 half specimens (40 mm x 40 mm x 80 mm) were used.

212 From an environmental point of view, the contaminating potential of EAFD must be an-  
213 alysed in the production of SCM, guaranteeing their safe use. For this reason, leaching  
214 tests were carried out, in monolithic and granular conditions, on the following mortars:  
215 SCM-EAFD1-25, SCM-EAFD1-100, SCM-EAFD2-25 and SCM-EAFD2-100.

216 The specimens for leaching tests were protected from water and moisture with plastic  
217 bags during the curing time so as not to affect the elements' mobility. The leaching tests  
218 in monolithic and granular state were carried out in accordance with XP X31-211:2012  
219 (AFNOR, 2017) and the compliance test UNE-EN 12457-4: 2003 (AENOR, 2017), re-  
220 spectively, carrying out two repetitions of each test. To perform the compliance test, the  
221 specimens were crushed and sieved previously. After the contact time of each test, an  
222 eluate was analysed by ICP-MS.

223 Regarding the leaching limits of monolithic waste, Member States are responsible for  
224 establishing the criteria to classify the material as inert, non-hazardous or hazardous.  
225 However, while the monolithic European limits are established, the granular limits can  
226 be applied, provided that the sample is crushed in advance. Due to the lack of European  
227 monolithic limits, some European countries, such as Germany, Finland, Sweden or Den-  
228 mark, apply the crushing of monolithic waste and its evaluation according to conventional  
229 leaching tests. However, the English Environmental Agency (EEA, 2010) uses the limits  
230 established in the European Directive 2003/33/CE (2003) for the monolithic waste veri-  
231 fication without prior crushing.

232 In this study, all the leaching results obtained, in a monolithic and granular state, were  
233 compared with the legal limits established by the European Directive 2003/33/CE (2003),  
234 shown in Table 1.

### 235 3. Results and discussion

#### 236 3.1. Materials characterisation

237 Table 4 shows the EAFD chemical composition results in oxides. The major components  
238 in both wastes were Zn and Fe at 38.27 and 29.88 for EAFD1 and 34.22 and 31.83 for  
239 EAFD2, respectively. The results agree with those of Vargas et al. (2006), López and  
240 López-Delgado (2002), Oustadakis et al. (2010) and Sapiña et al. (2014). The composi-  
241 tion of these wastes strongly depended on the steel manufactured and a slight change in  
242 the furnace may lead to a different EAFD composition even within the same steelworks.  
243 The presence of heavy metals (Zn, Pb, Cr) affects the cement setting and jeopardize the  
244 mortar properties (Lasheras-Zubiate et al., 2011; Tashiro and Oba, 1979). Regardless of  
245 the EAFD composition, components must be able to be released in the environment,  
246 which can be determined by leaching tests. The pH value is a decisive factor (Ledesma et  
247 al., 2018). There are no previous studies of how heavy metals affect the self-compacting  
248 properties in mortars.

249 The chemical composition results of CEM, SF, NS-0/2 and NS-0/4, expressed in oxides  
250 content (%), are shown in Table 5. The main component of the CEM was calcium oxide  
251 (CaO) with lower contents of silica oxide (SiO<sub>2</sub>), iron oxide (Fe<sub>2</sub>O<sub>3</sub>) or sulphur oxide  
252 (SO<sub>3</sub>), while SF was 100% silica oxide (SiO<sub>2</sub>). NS-0/2 and NS-0/4 had a siliceous char-  
253 acter mainly with minority contents of aluminium oxide (Al<sub>2</sub>O<sub>3</sub>) and potassium oxide  
254 (K<sub>2</sub>O) and iron oxide (Fe<sub>2</sub>O<sub>3</sub>).

255 Figure 1 shows the XRD patterns of both EAFD. The waste matrixes were composed  
256 mainly of Franklinite (ZnFe<sub>2</sub>O<sub>4</sub>) (22-1012) and Zincite (ZnO) (36-1451) (JCPD, 1995).  
257 These phases were also the main ones detected in the work carried out by da Silva

258 Magalhães et al. (2017). To a lesser extent, other phases were also detected, such as me-  
259 tallic Mn (33-0887), manganese oxide (MnO) (07-0230), quartz (SiO<sub>2</sub>) (33-1161) and  
260 lead oxide (PbO<sub>2</sub>) (37-0517) (JCPD, 1995).

261 The FTIR of EAFD1 and EAFD2 is shown in Figure 2. In these wastes, stretching and  
262 bending vibration characteristic bands of the OH<sup>-</sup> groups were observed (3504 and 1629  
263 cm<sup>-1</sup> in EAFD1 and 3643 and 1632 cm<sup>-1</sup> in EAFD2, respectively) (Farmer, 1974). The  
264 1170 and 1028 cm<sup>-1</sup> bands were associated to asymmetric stretching vibration of Si-O-Si,  
265 the 799 cm<sup>-1</sup> band in EAFD1 to symmetric stretching vibration of Si-O-Si and the 453 or  
266 427 cm<sup>-1</sup> bands to bending vibration of O-Si-O. The 577 cm<sup>-1</sup> band was associated to  
267 bending vibration O-Al-O (Calvo et al., 2013; Farinha et al., 2018; Martins et al., 2008).  
268 The presence of sulphates could be due to the 1087 cm<sup>-1</sup> band (Nakamoto, 1986). In  
269 EAFD1 and EAFD2 the presence of carbonates could be attributed to the 1425 cm<sup>-1</sup> and  
270 877 cm<sup>-1</sup> bands, associated to asymmetric stretching vibration of CO<sub>3</sub><sup>2-</sup> and bending out  
271 of plane, respectively (Calvo et al., 2013). . It is usual that the FTIR technique detects the  
272 presence of carbonate anion at low concentration levels, whose crystalline phases are not  
273 detected by XRD. Table 4 shows the presence of calcium oxide, so it is not surprising the  
274 presence of calcium carbonate, since calcium oxide hydrates and carbonates easily in the  
275 presence of ambient humidity.

276 Table 6 shows the specific surface area, specific gravity and water absorption values for the  
277 constituent materials of SCM. The specific surface area values of EAFD1 and EAFD2 were  
278 3.70 m<sup>2</sup>/g and 4.60 m<sup>2</sup>/g, respectively. Law et al. (1983) reported that the EAFD specific  
279 surface area was between 2.5 and 4 m<sup>2</sup>/g using the BET method. The specific gravity was  
280 3847 kg/m<sup>3</sup> for EAFD1 and 3809 kg/m<sup>3</sup> for EAFD2, in agreement with the results from de  
281 Vargas et al. (2006). The specific surface area and specific gravity of CEM was 0.35 m<sup>2</sup>/g

282 and 3140 kg/m<sup>3</sup>, respectively. For SF, the specific surface area was 0.25 m<sup>2</sup>/g and the spe-  
283 cific gravity was 2600 kg/m<sup>3</sup>. The specific gravity and water absorption of aggregates was  
284 2537 kg/m<sup>3</sup> and 0.41% for NS-0/2 and 2622 kg/m<sup>3</sup> and 0.46% for NS-0/4.

285 Figure 3 shows the particle size distribution of EAFD, SF and CEM, in distribution per-  
286 centage. CEM and SF exhibited a more homogeneous size distribution, whose values os-  
287 cillate between 0.06 and 100 µm, with a majority size of 20 µm in both materials. EAFD1  
288 and EAFD2 show bimodal curves with two major size fractions between 0.05-1 µm and  
289 1-20 µm, with maximum located in 0.3 µm in both wastes and in 2.5 µm for EAFD2 and  
290 in 3 µm for EAFD1. This distribution was in agreement with that reported by authors as  
291 Oustadakis et al. (2010). These authors detected mainly two fractions of size: one fine  
292 (0.1-1 µm) and one thicker (1-100 µm). Figure 4 shows the particles size distribution of  
293 NS-0/2 and NS-0/4, in mass, where the maximum size was 2 and 4 mm for NS-0/2 and  
294 NS-0/4, respectively.

295 Figures 5 and 6 show the EAFD SEM. Both figures show a typical spherical particle shape  
296 of EAFD (Dutra et al., 2006) forming groups around 10-15 µm (Lenz and Martins, 2007).

297 Regarding leaching behaviour of both wastes, Figure 7 shows the concentration released,  
298 expressed in mg/kg, for each element according to the compliance test UNE-EN 12457-  
299 4: 2003 (AENOR, 2017). The parameters of conductivity (µS/cm), temperature (°C) and  
300 pH measured during the test were also reflected. Comparing the metal concentration val-  
301 ues with the limit "Not hazardous" of European Directive 2003/33/CE (2003) of ac-  
302 ceptance of waste in landfill (plotted in blue in Figure 7). It was observed that Se and Cd  
303 exceeded the limit in EAFD1, while in EAFD2 it was Se, Mo and Pb. Due to this fact,  
304 both wastes were classified as "Hazardous".

305 The Pb concentration in EAFD2 released into the medium was much higher (5483.87

306 mg/kg) than the Pb release in EAFD1 (6.14 m/g/kg). This phenomenon could be explained  
307 by the higher pH of this EAFD2 (13.3) compared to the EAFD1 (9.6) registered during the  
308 compliance test made according to UNE-EN 12457-4:2003 (AENOR, 2017). At a pH  
309 higher than 12, Pb is mobilised easily (Pereira et al., 2001; Sebag et al., 2009; van der Sloot  
310 and Dijkstra, 2004). Ledesma et al. (2018) studied the pH influence on the leaching of  
311 EAFD. These authors revealed that the release of Pb was very low for pH values between  
312 9 and 11. This justifies the low release of Pb in the EAFD1 and the high release in EAFD2.

### 313 **3.2. Self-compactability fresh parameters**

314 Table 7 shows the self-compactability parameters used in SCRM, SCM-EAFD1 and  
315 SCM-EAFD2. In SCRM, the ratios  $V_w/V_p$  and  $S_p/p\%$  were established at 0.84 and 0.60 as  
316 the optimal self-compactability parameters. The needs for water and  $S_p$  were higher when  
317 EAFD was incorporated in the mixes. Properties such as fluidity, cohesion and the pass-  
318 ing ability in SCM are very sensitive to changes in the free water content, the source of  
319 the aggregate and the superplasticiser content (Kwan and Ling, 2017). For this reason,  
320 the ratios  $V_w/V_p$  and  $S_p/p\%$  were modified with respect to SCRM to guarantee self-com-  
321 pactability. EAFD was finer (Figure 3) and with greater specific surface (Table 6) waste  
322 than SF, thus the water and  $S_p$  needs were higher than for SCRM.

323 Comparing the behaviour of mortars with EAFD1 and EAFD2, it is observed that the incor-  
324 poration of EAFD1 requires a smaller amount of water ( $V_w/V_p$ ) and superplasticiser ( $S_p/p\%$ )  
325 than EAFD2 to achieve the parameters of self compactability (Table 7). This trend could be  
326 attributed to the particle size distribution of both wastes. In Figure 3, the largest volume of  
327 particles enclosed between the curve and the range 0.05-1  $\mu\text{m}$  corresponded to EAFD2  
328 (50.92%, *versus* 41.52% in EAFD1). The greater number of fine particles in EAFD2 could  
329 have caused a greater surface to hydrate. In addition, the greater specific surface area of

330 EAFD2 with respect to EAFD1 (Table 6) may have led to this behaviour.

331 Regarding  $G_m$  values, a decrease was observed as EAFD content increased, despite the  
332 fact that the self-compactability parameters ( $V_w/V_p$  and  $S_p/p\%$ ) increased. The values of  
333  $G_m$  obtained, setting the ratios  $V_w/V_p$  and  $S_p/p\%$ , were admissible due to none of the mixes  
334 showing segregation of aggregate or water exudation visual signs.

### 335 **3.3. Mechanical strength**

336 The compressive and flexural strengths of mortars containing EAFD were evaluated at 7,  
337 28 and 56 days of age according to UNE-EN 1015-11: 2000 (AENOR, 2017). The com-  
338 pressive and flexural strengths results of SCM-EAFD1 are shown in Figures 8 and 9,  
339 respectively. An abrupt loss of mechanical strength was observed with the incorporation  
340 of EAFD1 at 7 days relative to SCRM. This loss cannot be attributed to the w/c ratio since  
341 the variation was not significant when EAFD1 was incorporated (Table 7).

342 Figures 8 and 9 showed that SCM-EAFD1-25 increased mechanical strengths with values  
343 of 0.98, 22.39 and 48.17 MPa for compressive strength and 0.15, 0.85 and 1.71 MPa for  
344 flexural strength, at 7, 28 and 56 days, respectively. This phenomenon indicates a delay in  
345 reaching mechanical strength relative to SCRM. Numerous studies have corroborated the  
346 influence of heavy metals (Cu, Cr, Pb, Zn, Li, Hg) on the hydration and microstructure of  
347 cement (Lasheras-Zubiate et al., 2011; Tashiro and Oba, 1979). Generally, the delay that  
348 EAFD causes in the hydration of cement is attributed to the high concentration of Zn in the  
349 composition of this waste (Table 4) (Balderas et al., 2001; Fares et al., 2016). Specifically,  
350 the formation of a coating that envelops the cement particles, avoiding common hydration  
351 (Yousuf et al., 1995). However, there are discrepancies about which Zn phase is responsible  
352 for the coating. According to Yousuf et al. (1995), in solutions with sufficient presence of  
353  $Ca^{2+}$  ions, the different forms of zinc hydroxide ( $Zn(OH)_3^-$  and  $Zn(OH)_4^{2-}$ ) are transformed

354 into Calcium Zincate ( $\text{CaZn}_2(\text{OH})_6 \cdot 2\text{H}_2\text{O}$ ), which completely covers cement grains, slow-  
355 ing down the hydration reactions. However, Castellote et al. (2004), de Vargas et al. (2006)  
356 and Brehm et al. (2017) state that it is Calcium Zincate ( $\text{CaZn}_2(\text{OH})_6 \cdot 2\text{H}_2\text{O}$ ) that allows the  
357 hydration of the paste to restart after the corresponding delay.

358 Castellote et al. (2004) observed a similar behaviour, where the reference mortar had a com-  
359 pressive strength of 25 MPa and 38 MPa at 7 and 28 days, respectively. Mortars with an  
360 EAFD/cement ratio equal to 0.86 did not gain compressive strength until 12 days, when it  
361 was less than 3 MPa. However, at 28 days these mortars reached approximately 35 MPa. In  
362 our study, SCM-EAFD1-25 has a lower EAFD/cement ratio (Table 3) than the one used by  
363 Castellote et al. (2004) and compressive strengths lower than 35 MPa at 28 days.

364 The mechanical strength evolution of the mortars with 50% and 100% of EAFD1 was  
365 different from that of those with 25%. The compressive and flexural strengths values rec-  
366 orded by SCM-EAFD1-50 and SCM-EAFD1-100 were very similar at all ages of curing.  
367 This phenomenon was related to the higher content of EAFD1 in mortars. The metal con-  
368 tent in 50% and 100% replacement could cause the hydration reactions not to happen  
369 before 56 days of curing. For SCM-EAFD1 mortars, a replacement percentage of SF with  
370 EAFD1 greater than 25% is not viable from a mechanical point of view.

371 Although the compressive strength was considered insufficient for SCM-EAFD1-50 and  
372 SCM-EAFD1-100, the values recorded at all ages exceeded the limits set by the English En-  
373 vironmental Agency (EEA, 2010) and XP X31-211: 2012 (AFNOR, 2017) to be considered  
374 as monolithic material deposited in landfill, although this is not the focus of the study.

375 Figures 10 and 11 show the mechanical strengths of SCM-EAFD2. In both figures an  
376 adverse effect of EAFD2 was observed in the development of compressive and flexural  
377 strengths, since these decrease with the higher EAFD2 content in mortars. This decrease



378 could be related to the following isolated and/or combined events: i) the higher EAFD2  
379 content and consequently greater heavy metals presence in mortars that interfere in me-  
380chanical strength development; ii) the greater w/c ratio as a consequence of the greater  
381 volume of fine particles and the high specific surface area of EAFD2.

382 Previous works by Ledesma et al. (2017) recorded lower compressive strengths compared  
383 to the reference mortar, when the EAFD/cement ratio increased in cylindrical specimens of  
384 conventional mortar made with limestone filler and aggregate. This behaviour was also  
385 attributed to the greater presence of heavy metals that interfered in the common develop-  
386 ment of mechanical strengths and greater need for water in mortars with EAFD. In the  
387 research by da Silva Magalhães et al. (2017), the mechanical strength, with 5% and 10% of  
388 EAFD as cement replacement, decreased with the increase in the w/c ratio.

389 All mortars with EAFD2 were viable from a mechanical point of view. Despite mechan-  
390 ical strength reduction in SCM-EAFD2, the values of the SCM-EAFD2-25 mortar were  
391 very close to those of SCRM at 56 days. These losses only represented 1.4%.

392 Mechanical strengths increased in all mortars as the curing age increased (mechanical  
393 strength -  $MS_{7d} < MS_{28d} < MS_{56d}$ ).

394 Figure 12 shows linear relationships of flexural and compressive strengths obtained by  
395 the mortars at 7, 28 and 56 days, with good  $R^2$  values. The flexural strength increases to  
396 a lesser extent compared to the compressive strength at 56 days, which is consistent with  
397 the generalized tendency for conventional concrete (Neville, 1995). This phenomenon  
398 was also observed by da Silva Magalhães et al. (2017) when they evaluated the cementing  
399 efficiency of EAFD in mortars.

### 400 3.4. Study of mineralogical phases

401 Figure 13 shows the mineral phases formed in SCRM during curing. The main detected  
402 phases correspond to Quartz ( $\text{SiO}_2$ ) (33-1161) and Portlandite ( $\text{Ca(OH)}_2$ ) (04-0733)  
403 (JCPD, 1995). The presence of Microcline ( $\text{KAlSi}_3\text{O}_6$ ) (19-0932), Calcite ( $\text{CaCO}_3$ ) (05-  
404 0586), dicalcium silicate ( $\text{C}_2\text{S}$ ) (24-0034) and tricalcium silicate ( $\text{C}_3\text{S}$ ) (42-0551) (JCPD,  
405 1995) was also detected. At 56 days of curing, slight diffraction peaks associated to  
406 Ettringite ( $\text{Ca}_6\text{Al}_2(\text{SO}_4)_3(\text{OH})_{12}\cdot 26\text{H}_2\text{O}$ ) (41-1451) (JCPD, 1995) were also observed.  
407 This phase is formed during the cement hydration and, under normal curing conditions,  
408 it does not produce expansion (Skalny et al., 1996).

409 The X-ray diffraction patterns of hardened mortars containing EAFD are shown in Fig-  
410 ures 14-19. As in SCRM, the prevailing phase was Quartz ( $\text{SiO}_2$ ) (33-1161) (JCPD,  
411 1995), due to the siliceous character of both the sands and filler used. Additionally, Frank-  
412 linite ( $\text{ZnFe}_2\text{O}_4$ ) (22-1012) and Zincite ( $\text{ZnO}$ ) (36-1451) phases (JCPD, 1995), due to the  
413 composition of EAFD, were detected since diffraction peaks relative to these phases were  
414 detected in the wastes in their original state (Figure 1). A Cassiterite phase ( $\text{SnO}_2$ ) (41-  
415 1445) (JCPD, 1995) was observed in the mortars containing EAFD1, due to the greater  
416 presence of Sn in their chemical composition. In mortars with 100% (Figures 16 and 19)  
417 and 50% (Figure 15) incorporation of EAFD1, Calcium Zincate ( $\text{CaZn}_2(\text{OH})_6\cdot 2\text{H}_2\text{O}$ ) (25-  
418 1449) (JCPD, 1995) was detected, a characteristic phase of the reaction of mortar with  
419 EAFD (Castellote et al., 2004; de Vargas et al., 2006; Ledesma et al., 2017). Other phases,  
420 such as Microcline ( $\text{KAlSi}_3\text{O}_6$ ) (19-0932), Orthoclase ( $\text{KAlSi}_3\text{O}_6$ ) (31-0966), Calcite  
421 ( $\text{CaCO}_3$ ) (05-0586), dicalcium silicate ( $\text{C}_2\text{S}$ ) (24-0034) and tricalcium silicate ( $\text{C}_3\text{S}$ ) (42-  
422 0551), Ettringite ( $\text{Ca}_6\text{Al}_2(\text{SO}_4)_3(\text{OH})_{12}\cdot 26\text{H}_2\text{O}$ ) (41-1451) and Iron sulphate hydroxide  
423 ( $\text{Fe}_4(\text{OH})_{10}\text{SO}_4/2\text{Fe}_2\text{O}_3\cdot \text{SO}_3\cdot 5\text{H}_2\text{O}$ ) (21-0429) (JCPD, 1995) were also detected.

424 Diffraction peaks associated with a Portlandite phase ( $\text{Ca(OH)}_2$ ) (04-0733) (JCPD, 1995)

425 were observed in the SCM-EAFD1-25 mortars at 28 and 56 days (Figure 14). Additionally,  
426 this phase also appeared in all mortars SCM-EAFD2 at 7, 28 and 56 days (Figures 17-19).  
427 However, the Portlandite phase ( $\text{Ca}(\text{OH})_2$ ) (04-0733) (JCPD, 1995) was not detected in  
428 SCM-EAFD1-25 at 7 days (Figure 14), neither in SCM-EAFD1-50 (Figure 15) nor in  
429 SCM-EAFD1-100 at any age (Figure 16). Therefore, the mechanical strength achieved  
430 could be related to the presence of Portlandite ( $\text{Ca}(\text{OH})_2$ ) (04-0733) (JCPD, 1995). The  
431 amount of Portlandite ( $\text{Ca}(\text{OH})_2$ ) identified by DRX analysis is indicative of the extension  
432 of the Portland cement reaction (Ledesma et al., 2017). The SCM-EAFD1-25 mortar's  
433 XRD patterns (Figure 14) show that Portlandite ( $\text{Ca}(\text{OH})_2$ ) does not manifest itself after  
434 7 days; however, it was detected at 28 and 56 days, due to the mechanical behaviour  
435 previously explained. The addition of heavy metals, such as Cd, Ni or Pb to the paste  
436 causes a decrease in the content of Portlandite ( $\text{Ca}(\text{OH})_2$ ), increasing their weakness  
437 (Gollmann et al., 2010). In the XRD research of de Vargas et al. (2006), it was observed  
438 that the content of Portlandite ( $\text{Ca}(\text{OH})_2$ ) decreases with the higher content of EAFD.

439 Figures 15 and 16 show the mineralogical phases of SCM-EAFD1-50 and SCM-EAFD1-  
440 100, respectively. In these mortars, Portlandite ( $\text{Ca}(\text{OH})_2$ ) was not observed; but Calcium  
441 Zincate ( $\text{CaZn}_2(\text{OH})_6 \cdot 2\text{H}_2\text{O}$ ) was. Ledesma et al. (2017) attributed the loss of mechanical  
442 strengths of EAFD mortars EAFD to the formation of this phase ( $\text{CaZn}_2(\text{OH})_6 \cdot 2\text{H}_2\text{O}$ ) that  
443 prevented the formation of Portlandite ( $\text{Ca}(\text{OH})_2$ ). In our research, the mechanical behaviour  
444 of SCM-EAFD1-50 and SCM-EAFD1-100 mortars could be related to the appearance of  
445 Calcium Zincate ( $\text{CaZn}_2(\text{OH})_6 \cdot 2\text{H}_2\text{O}$ ) as indicated by Ledesma et al. (2017). In mortars in  
446 which Portlandite ( $\text{Ca}(\text{OH})_2$ ) was not detected, hydration may not have started and/or ended  
447 because of EAFD presence. Conversely, regarding SCM-EAFD2-100, Figure 19 showed  
448 joint formation of Portlandite ( $\text{Ca}(\text{OH})_2$ ) and Calcium Zincate ( $\text{CaZn}_2(\text{OH})_6 \cdot 2\text{H}_2\text{O}$ ), guaran-  
449 teeing that both phases can be formed together during the hydration process, as stated by

450 Castellote et al. (2004), de Vargas et al. (2006) and Brehm et al. (2017).

451 Additionally (Figures 14-19), as the content of EAFD increases in the mortars, the content of  
452 tricalcium silicate ( $C_3S$ ) increases slightly for the same hydration age, indicating in turn a  
453 lower presence of Portlandite ( $Ca(OH)_2$ ) and worse mechanical behaviour. For the same mor-  
454 tar with the same EAFD amount, it was observed that tricalcium silicate decreases ( $C_3S$ ) and  
455 Portlandite increases ( $Ca(OH)_2$ ), with curing time, which will be the more pronounced the  
456 lower the EAFD content. Balderas et al. (2001) carried out a study on tricalcium silicate ( $C_3S$ )  
457 phase consumption in a cementitious paste doped with 10% EAFD. In this research, they  
458 observed that the EAFD paste consumed less tricalcium silicate ( $C_3S$ ) than the undoped mor-  
459 tar, having a lower hydration reaction and worse final mechanical strength.

### 460 **3.5. Water absorption by capillarity**

461 The water absorption by capillarity results in accordance with UNE-EN 1015-18:2003  
462 (AENOR, 2017) at 28 days are shown in Table 8. SCRM showed a water absorption value  
463 by capillarity of  $0.095 \text{ kg}/(\text{m}^2 \cdot \text{min}^{0.5})$ .

464 The water absorption by capillarity underwent a large increase when EAFD1 was incor-  
465 porated into mortars, reaching a maximum value with SCM-EAFD1-50 of  $1,077$   
466  $\text{kg}/(\text{m}^2 \cdot \text{min}^{0.5})$ . Regarding 50% and 100% replacement of SF by EAFD1, the mechanical  
467 strength was not achieved, because of an incorrect and/or delay cement setting time, stop-  
468 ping the formation of a cementitious matrix. High capillary coefficients represent high  
469 absorption velocity values, which is undesirable from the point of view of mechanical  
470 and leaching behaviour (Lucas et al., 2016). Therefore, the results of the SCM-EAFD1  
471 mortars dissuade the use of a percentage higher than 25% replacement of SF.

472 In the SCM-EAFD2 mortars, capillarity increases as the EAFD2 content increased, ob-  
473 taining a maximum value for the SCM-EAFD2-100 mortar with  $0.209 \text{ kg}/(\text{m}^2 \cdot \text{min}^{0.5})$ .

474 Previous EAFD management studies have not taken this property into account. Figure 20  
475 shows the relationship between capillarity and compressive strengths of mortars. It was  
476 observed that mortars with better mechanical behaviour have less water absorption by  
477 capillarity since there was a lower interconnection in the pores. This increase could be  
478 related to the increase in the w/c ratio when incorporating EAFD2 (Table 7).

### 479 **3.6. Leaching behaviour**

480 The element release results according to XP X31-211: 2012 (AFNOR, 2017) of the mor-  
481 tars SCM-EAFD1-25, SCM-EAFD1-100, SCM-EAFD2-25 and SCM-EAFD2-100 are  
482 shown in Figures 21 and 22, respectively. In addition, the conductivity ( $\mu\text{S}/\text{cm}$ ), temper-  
483 ature ( $^{\circ}\text{C}$ ) and pH recorded during the test are shown. The "Non-Hazardous" limit of the  
484 European Waste Landfill Acceptance Directive (in blue) is also displayed. In mortars  
485 SCM-EAFD1-25, SCM-EAFD1-100, SCM-EAFD2-25 and SCM-EAFD2-100, none of  
486 the elements released into the environment exceeded the "Non-hazardous" limit estab-  
487 lished by the European Directive (2003).

488 Ledesma et al. (2017) carried out a mechanical stabilisation study of EAFD using two  
489 conventional mortars. In that study, mortar (cement and aggregate) was replaced by  
490 EAFD with a ratio of EAFD/CEM of 3.33 and 6.67. Although the authors reduced the Pb  
491 released in approximately 98%, this release exceeded the limit of the European Directive  
492 (2003) and none of the mortars could be classified as "Non-Hazardous". The authors at-  
493 tributed the non-stabilisation of EAFD to the basic pH registered during the monolithic  
494 leaching test. Our research succeeded in stabilising a maximum amount of 573.9 kg and  
495 493.2 kg of EAFD/ $\text{m}^3$  of mortar for EAFD1 and EAFD2, respectively, which is equiva-  
496 lent to an EAFD/CEM ratio of 1.22. In our study, the effectiveness in monolithic state  
497 resides in the higher density of the immobilization matrix. The SCM produced were  
498 denser (values around  $2.6 \text{ g}/\text{cm}^3$ ) than the conventional ones used by Ledesma et al.

499 (2017) (1.14 - 1.60 g/cm<sup>3</sup>), allowing a better EAFD encapsulation.

500 Figures 23 and 24 show the elements released of the mortars SCM-EAFD1-25, SCM-  
501 EAFD1-100, SCM-EAFD2-25 and SCM-EAFD2-100 in accordance with the compliance  
502 test UNE EN-12457-4: 2003 (AENOR, 2017). These figures show the conductivity  
503 ( $\mu\text{S}/\text{cm}$ ), temperature ( $^{\circ}\text{C}$ ) and pH measured during the test and the "Non-Hazardous"  
504 limit established in the European Directive (2003) (coloured blue).

505 The SCM-EAFD1-25, SCM-EAFD1-100 and SCM-EAFD2-100 mortars exceeded the  
506 "Non-Hazardous" limit of the European Directive (2003) due to the release of Pb with  
507 values of 13,431, 40,521 and 44,301 mg/kg. It was observed that Pb release was higher  
508 in mortars with greater content of EAFD.

509 The SCM-EAFD2-25 mortar was classified as "Non-Hazardous", because all elements  
510 were kept below the legal limit in granular state, allowing a safe use of this material.

511 Several authors reported difficulty in incorporating Pb treatment into cementitious mate-  
512 rials because the high alkalinity of cement makes its mobility easier (Ledesma et al., 2017;  
513 Navarro et al., 2011). The pH of the cement-based materials ranges from 12 to 13. These  
514 values coincide with the range of highest Pb release in the EAFD that is around 12  
515 (Ledesma et al., 2018).

516 For 25% of replacement in both wastes, the Pb mobility could be more related to the phys-  
517 ical characteristics of the matrix containing the waste. Figure 20 shows that mortars with  
518 higher compressive strength have a lower water absorption by capillarity, therefore being  
519 less accessible and susceptible to leaching. In this sense, the Pb in SCM-EAFD2-25 could  
520 maintain its leaching (9,017 mg/kg) lower than the allowed limit ( $< 10 \text{ mg}/\text{kg}$ ) due to the  
521 encapsulation matrix generated by the SCM, although its leaching was favourable from a

522 chemical point of view. The mobility of Pb is promoted in the alkaline environment pro-  
523 duced by cement (Laforest and Duchesne, 2007). This mortar has values of compression  
524 strength and water absorption by capillarity very close to those of SCRM (Figure 20). The  
525 SCM-EAFD1-25 mortar was located further to the left in Figure 20, indicating lower com-  
526 pression strength values and greater water absorption by capillarity, thus favouring a greater  
527 Pb mobility in this matrix.

528 Because of the mortars location in Figure 20, SCM-EAFD2-100 (44,301 mg/kg) should  
529 have less leaching than SCM-EAFD1-100 (40,521 mg/kg). However, when the percent-  
530 age of replacement of SF with EAFD was 100%, Pb leaching seems to be promoted by  
531 registered pH values and the greater amount of waste. Navarro et al. (2011) reported that  
532 the pH optimum to Pb precipitation was approximately 10. In the SCM-EAFD1-100 mor-  
533 tar (Figure 23), the difference between the pH of EAFD1 in the original state (9.6) and  
534 that reached through the compliance test (12.7) in this mortar could be responsible for the  
535 mobility of Pb. For SCM-EAFD2-100 (Figure 24) there was no significant difference  
536 between the pH of EAFD2 in the original state (13.3) and that reached during the com-  
537 pliance test (13.1). Therefore, this pH value encourages the leaching of Pb.

#### 538 **4. Conclusions**

539 This study shows the influence of the use of two samples of electric arc furnace dust  
540 (EAFD1 and EAFD2), as replacement of siliceous filler (SF), used in the production of  
541 SCM. The main conclusions were:

- 542 - Both EAFD showed similar characteristics, with predominance of Zn and Fe ox-  
543 ides and mineralogical phases Franklinite ( $ZnFe_2O_4$ ) and Zincite ( $ZnO$ ). The spe-  
544 cific surface area of EAFD2 ( $4.60 \text{ m}^2/\text{g}$ ) was higher than that of EAFD1 ( $3.70$   
545  $\text{m}^2/\text{g}$ ). The particle size distribution of both EAFD showed two fractions of sizes

546 and fine particles were higher in EAFD2. Regarding leaching behaviour, the waste  
547 was classified as "Hazardous" since elements such as Se and Cd exceeded the  
548 limit allowed in EAFD1, and Se, Mo and Pb in EAFD2. pH differences were  
549 found in EAFD1 (9.6) and EAFD2 (13.3). The higher pH value in EAFD2 pro-  
550 moted the high release of Pb that this waste registered;

551 - SCM production using EAFD as raw secondary material was possible. When in-  
552 corporating EAFD in the mortars, it was necessary to increase the amounts of  
553 water and superplasticiser, to guarantee the self-compacting properties, due to the  
554 higher content of fine particles and the bigger specific surface area of these wastes  
555 compared to SF;

556 - The mechanical strength decreased for higher EAFD content. From a mechanical  
557 point of view, more than 25% replacement of SF with EAFD1 is not  
558 recommended. Despite a delay in the gain of mechanical strength, the mortar with  
559 25% of EAFD1 showed a compressive strength of 48.17 MPa at 56 days. The  
560 mortars with EAFD2 showed lower strength losses, reaching 65.28 MPa of  
561 compressive strength at 56 days for 25% of EAFD2. This value was very close to  
562 that of the reference mortar at the same age (66.20 MPa);

563 - Characteristic phases of EAFD hardened mortars, such as Franklinite ( $ZnFe_2O_4$ ),  
564 Zincite ( $ZnO$ ) and Calcium Zincate ( $CaZn_2(OH)_6 \cdot 2H_2O$ ), were detected. The  
565 mechanical strength was related to Portlandite ( $Ca(OH)_2$ ) formation. This phase  
566 was not detected during the mineralogical study in the mortars with 50% and  
567 100% replacement of SF with EAFD1;

568 - Water absorption by capillarity increased with greater presence of EAFD, giving  
569 rise to self-compacting matrices with more interconnected pores, resulting in



570 worse mechanical strength and matrices more susceptible to leaching;

571 - The monolithic leaching test showed that all mortars were environmentally safe  
572 because the elements showed releases below the established limit (10 mg/kg).  
573 This was related to the SCM matrix that allows better EAFD encapsulation;

574 - Pb mobility in mortar with 25% of EAFD2 in the granular state was related to the  
575 physical characteristics of the self-compacting matrix. This mortar, due to its better  
576 mechanical strength and lower capillarity, showed lesser susceptibility to leaching  
577 relative to that with 25% of EAFD1 and it was the only one that was environmentally  
578 safe to use. In mortars with 100% of EAFD1 and EAFD2 replacement, Pb leaching  
579 was promoted by the pH values registered in the original state of the EAFD1 and  
580 EAFD2 and those reached during the compliance test in granular state.

581 The results indicate that the use of less than 25% of EAFD1 in SCM is possible if Pb  
582 release can be controlled in the granular state, because it is viable from a mechanical  
583 viewpoint. Regarding the mortar with 25% of EAFD2, the requirements on mechanical  
584 strengths and monolithic and granular leaching behaviour were achieved. EAFD2 up to  
585 25% is viable to produce SCM, above which mechanisms that control the release of Pb  
586 are necessary in the granular state. Through this research, a hazardous waste was safely  
587 managed through its incorporation in a construction process, such as the manufacture of  
588 SCM, thus promoting Circular Economy.

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### 603 **References**

- 604 AENOR, 2017. Asociación Española de normalización y certificación, AENOR, Madrid,  
605 Spain, [www.aenor.es](http://www.aenor.es).  
606
- 607 AFNOR, 2017. Association française de normalisation, [www.afnor.org](http://www.afnor.org).  
608
- 609 Anastasiou, E., Filikas, K.G., Stefanidou, M., 2014. Utilization of fine recycled  
610 aggregates in concrete with fly ash and steel slag. *Construction and Building Materials*  
611 *50*, 154-161.  
612
- 613 Balderas, A., Navarro, H., Flores-Velez, L.M., Dominguez, O., 2001. Properties of  
614 portland cement pastes incorporating nanometer-sized franklinite particles obtained from  
615 electric-arc-furnace dust. *Journal of the American Ceramic Society* *84*(12), 2909-2913.  
616
- 617 Brehm, F.A., Moraes, C.A.M., Modolo, R.C.E., Vilela, A.C.F., Dal Molin, D.C.C., 2017.  
618 Oxide zinc addition in cement paste aiming electric arc furnace dust (EAFD) recycling.  
619 *Construction and Building Materials* *139*, 172-182.  
620
- 621 Calvo, J.L.G., Moreno, M.S., Alonso, M.C.A., Lopez, A.H., Olmo, J.G., 2013. Study of  
622 the microstructure evolution of low-pH cements based on ordinary Portland cement  
623 (OPC) by mid- and near-infrared spectroscopy, and their influence on corrosion of steel  
624 reinforcement. *Materials* *6*(6), 2508-2521.  
625
- 626 Castellote, M., Menendez, E., Andrade, C., Zuloaga, P., Navarro, M., Ordonez, M., 2004.  
627 Radioactively contaminated electric arc furnace dust as an addition to the immobilization  
628 mortar in low- and medium-activity repositories. *Environmental Science & Technology*  
629 *38*(10), 2946-2952.  
630
- 631 Chen, Q.Y., Tyrer, M., Hills, C.D., Yang, X.M., Carey, P., 2009. Immobilisation of heavy  
632 metal in cement-based solidification/stabilisation: a review. *Waste Management* *29*(1),  
633 390-403.

634  
635 Cholake, S.T., Farzana, R., Numata, T., Sahajwalla, V., 2018. Transforming electric arc  
636 furnace waste into value added building products. *Journal of Cleaner Production* 171,  
637 1128-1139.  
638  
639 COM(2017)33, 2017. European Commission, report from the commission to the  
640 European parliament, the council, the European economic and social committee and the  
641 committee of the regions on the implementation of the Circular Economy action plan.  
642 Available from: [http://ec.europa.eu/environment/circular-](http://ec.europa.eu/environment/circular-economy/implementation_report.pdf)  
643 [economy/implementation\\_report.pdf](http://ec.europa.eu/environment/circular-economy/implementation_report.pdf) (accessed September, 2018).

644 da Silva Magalhães, M., Faleschini, F., Pellegrino, C., Brunelli, K., 2017. Cementing  
645 efficiency of electric arc furnace dust in mortars. *Construction and Building Materials*  
646 157 (Supplement C), 141-150.  
647  
648 da Silva, P.R., de Brito, J., 2015. Fresh-state properties of self-compacting mortar and  
649 concrete with combined use of limestone filler and fly ash. *Materials Research-Ibero-*  
650 *American Journal of Materials* 18(5), 1097-1108.  
651  
652 de Vargas, A.S., Masuero, A.B., Vilela, A.C.F., 2006. Investigations on the use of  
653 electric-arc furnace dust (EAFD) in pozzolan-modified Portland cement I (MP) pastes.  
654 *Cement and Concrete Research* 36(10), 1833-1841.  
655  
656 Dutra, A.J.B., Paiva, P.R.P., Tavares, L.M., 2006. Alkaline leaching of zinc from electric  
657 arc furnace steel dust. *Minerals Engineering* 19(5), 478-485.  
658  
659 EEA, 2010. English Environmental Agency. Available from:  
660 [https://www.gov.uk/government/uploads/system/uploads/attachment\\_data/file/296422/g](https://www.gov.uk/government/uploads/system/uploads/attachment_data/file/296422/gho1110btew-e-e.pdf)  
661 [ho1110btew-e-e.pdf](https://www.gov.uk/government/uploads/system/uploads/attachment_data/file/296422/gho1110btew-e-e.pdf) (accessed September, 2018).  
662  
663 EFNARC, 2005. The European guidelines for self-compacting concrete specification.  
664 Production and use. Hampshire, UK. Available from: [www.efnarc.org](http://www.efnarc.org) (accessed  
665 September, 2018).  
666  
667 Esquinas, A.R., Ramos, C., Jiménez, J.R., Fernández, J.M., de Brito, J., 2017. Mechanical  
668 behaviour of self-compacting concrete made with recovery filler from hot-mix asphalt  
669 plants. *Construction and Building Materials* 131, 114-128.  
670  
671 EU, 2003. European Council Decision for the acceptance of waste at landfills. Available  
672 from: [https://eur-](https://eur-lex.europa.eu/LexUriServ/LexUriServ.do?uri=OJ:L:2003:011:0027:0049:EN:PDF)  
673 [lex.europa.eu/LexUriServ/LexUriServ.do?uri=OJ:L:2003:011:0027:0049:EN:PDF](https://eur-lex.europa.eu/LexUriServ/LexUriServ.do?uri=OJ:L:2003:011:0027:0049:EN:PDF)  
674 (accessed September, 2018).  
675  
676 EWC, 2002. European Waste Catalog. Available from:  
677 <https://www.boe.es/boe/dias/1999/01/08/pdfs/A00570-00580.pdf> (accessed September, 2018).  
678  
679 Fares, G., Al-Zaid, R.Z., Fauzi, A., Alhozaimy, A.M., Al-Negheimish, A.I., Khan, M.I.,  
680 2016. Performance of optimized electric arc furnace dust-based cementitious matrix  
681 compared to conventional supplementary cementitious materials. *Construction and*  
682 *Building Materials* 112, 210-221.  
683

684 Farinha, C.B., de Brito, J., Veiga, R., Fernández, J., Jiménez, J., Esquinas, A., 2018.  
685 Wastes as aggregates, binders or additions in mortars: selecting their role based on  
686 characterization. *Materials* 11(3), 453.  
687  
688 Farmer, V.C., 1974. *The Infrared spectra of minerals*. Mineralogical Society, London.  
689  
690 Gollmann, M.A.C., da Silva, M.M., Masuero, A.B., dos Santos, J.H.Z., 2010.  
691 Stabilization and solidification of Pb in cement matrices. *Journal of Hazardous Materials*  
692 179(1-3), 507-514.  
693  
694 Issa, H., Korac, M., Gavrilovski, M., Pavlovic, M., Kamberovic, Z., 2012. Possibility of  
695 carbon steel EAFD solidification/stabilization in concrete. *Revista De Chimie* 63(10),  
696 1008-1012.  
697  
698 JCPD, 1995. *Joint Committee on Powder Diffraction Standard-International Centre for*  
699 *Diffraction Data*. Swarthmore, PA.  
700  
701 Kanadasan, J., Razak, H.A., Subramaniam, V., 2018. Properties of high flowable mortar  
702 containing high volume palm oil clinker (POC) fine for eco-friendly construction. *Journal*  
703 *of Cleaner Production* 170, 1244-1259.  
704  
705 Kelebek, S., Yörük, S., & Davis, B. (2004). Characterization of basic oxygen furnace dust  
706 and zinc removal by acid leaching. *Minerals Engineering*, 17(2), 285-291.  
707  
708 Khotbehsara, M.M., Miyandehi, B.M., Naseri, F., Ozbakkaloglu, T., Jafari, F., Mohseni,  
709 E., 2018. Effect of SnO<sub>2</sub>, ZrO<sub>2</sub>, and CaCO<sub>3</sub> nanoparticles on water transport and durability  
710 properties of self-compacting mortar containing fly ash: experimental observations and  
711 ANFIS predictions. *Construction and Building Materials* 158, 823-834.  
712  
713 Kim, Y., Hanif, A., Usman, M., Munir, M.J., Kazmi, S.M.S., Kim, S., 2018. Slag waste  
714 incorporation in high early strength concrete as cement replacement: environmental  
715 impact and influence on hydration & durability attributes. *Journal of Cleaner Production*  
716 172, 3056-3065.  
717  
718 Kwan, A.K.H., Ling, S.K., 2017. Filler technology for improving robustness and reducing  
719 cementitious paste volume of SCC. *Construction and Building Materials* 153  
720 (Supplement C), 875-885.  
721  
722 Laforest, G., Duchesne, J., 2006. Stabilization of electric arc furnace dust by the use of  
723 cementitious materials: ionic competition and long-term leachability. *Cement and*  
724 *Concrete Research* 36(9), 1628-1634.  
725  
726 Laforest, G., Duchesne, J., 2007. Investigation of stabilization/solidification for treatment  
727 of electric arc furnace dust: dynamic leaching of monolithic specimens. *Cement and*  
728 *Concrete Research* 37(12), 1639-1646.  
729  
730 Lasheras-Zubiate, M., Navarro-Blasco, I., Álvarez, J.I., Fernández, J.M., 2011.  
731 Interaction of carboxymethylchitosan and heavy metals in cement media. *Journal of*  
732 *Hazardous Materials* 194(Supplement C), 223-231.  
733

734 Law, S.L., Lowry, W., Snyder, J., Kramer, G., 1983. Characterization of steelmaking  
735 dusts from electric arc furnaces. U.S. Dep. Interior, Bur. Mines.  
736

737 Ledesma, E.F., Lozano-Lunar, A., Fernández, J.M., Ayuso, J., Galvín, A.P., Jiménez,  
738 J.R., 2018. The role of pH on leaching of heavy metals and chlorides from Electric Arc  
739 Furnace Dust in cement-based mortars. *Construction and Building Materials* 183, 365-  
740 375.  
741

742 Ledesma, E.F., Jimenez, J.R., Ayuso, J., Fernandez, J.M., de Brito, J., 2017. Experimental  
743 study of the mechanical stabilization of electric arc furnace dust using fluid cement  
744 mortars. *Journal of Hazardous Materials* 326, 26-35.  
745

746 Lenz, D.M., Martins, F.B., 2007. Lead and zinc selective precipitation from leach electric  
747 arc furnace dust solutions. *Matéria (Rio de Janeiro)* 12(3), 503-509.  
748

749 Lucas, J., de Brito, J., Veiga, R., Farinha, C., 2016. The effect of using sanitary ware as  
750 aggregates on rendering mortars' performance. *Materials & Design* 91 (Supplement C),  
751 155-164.  
752

753 López, F., López-Delgado, A., 2002. Enhancement of electric arc furnace dust by recycling  
754 to electric arc furnace. *Journal of Environmental Engineering* 128(12), 1169-1174.  
755

756 López-Uceda, A., Ayuso, J., Jiménez, J.R., Agrela, F., Barbudo, A., De Brito, J., 2016.  
757 Upscaling the use of mixed recycled aggregates in non-structural low cement concrete.  
758 *Materials* 9(2), 91.  
759

760 Martins, F.M., Neto, J.M.D., da Cunha, C.J., 2008. Mineral phases of weathered and  
761 recent electric arc furnace dust. *Journal of Hazardous Materials* 154(1-3), 417-425.  
762

763 Maslehuddin, M., Awan, F.R., Shameem, M., Ibrahim, M., Ali, M.R., 2011. Effect of  
764 electric arc furnace dust on the properties of OPC and blended cement concretes.  
765 *Construction and Building Materials* 25(1), 308-312.  
766

767 Nakamoto, K., 1986. Infrared and raman spectra of inorganic and coordination  
768 compounds, Wiley.  
769

770 Navarro, A., Cardellach, E., Corbella, M., 2011. Immobilization of Cu, Pb and Zn in  
771 mine-contaminated soils using reactive materials. *Journal of Hazardous Materials* 186(2-  
772 3), 1576-1585.  
773

774 Nepomuceno, M., 2005. Methodology for self-compacting concrete mix design (in  
775 Portuguese). PhD thesis in civil engineering, University of Beira Interior, Portugal, 799 p.

776 Nepomuceno, M., Oliveira, L., Lopes, S.M.R., 2012. Methodology for mix design of the  
777 mortar phase of self-compacting concrete using different mineral additions in binary  
778 blends of powders. *Construction and Building Materials* 26(1), 317-326.  
779

780 Neville, A.M., 1995. Properties of concrete. Longman London.  
781

782 Nyirenda, R.L., 1991. The processing of steelmaking flue-dust: A review. *Minerals*  
783 *Engineering* 4(7), 1003-1025.

784  
785 Oustadakis, P., Tsakiridis, P.E., Katsiapi, A., Agatzini-Leonardou, S., 2010.  
786 Hydrometallurgical process for zinc recovery from electric arc furnace dust (EAFD): Part  
787 I: Characterization and leaching by diluted sulphuric acid. *Journal of Hazardous Materials*  
788 179(1), 1-7.  
789  
790 Pereira, C.F., Rodriguez-Pinero, M., Vale, J., 2001. Solidification/stabilization of electric  
791 arc furnace dust using coal fly ash - Analysis of the stabilization process. *Journal of*  
792 *Hazardous Materials* 82(2), 183-195.  
793  
794 Rattanashotinunt, C., Tangchirapat, W., Jaturapitakkul, C., Cheewaket, T., Chindaprasirt,  
795 P., 2018. Investigation on the strength, chloride migration, and water permeability of eco-  
796 friendly concretes from industrial by-product materials. *Journal of Cleaner Production*  
797 172, 1691-1698.  
798  
799 Salihoglu, G., Pinarli, V., 2008. Steel foundry electric arc furnace dust management:  
800 stabilization by using lime and Portland cement. *Journal of Hazardous Materials* 153(3),  
801 1110-1116.  
802  
803 Santamaría, A., Orbe, A., Losañez, M.M., Skaf, M., Ortega-Lopez, V., González, J.J.,  
804 2017. Self-compacting concrete incorporating electric arc-furnace steelmaking slag as  
805 aggregate. *Materials & Design* 115, 179-193.  
806  
807 Sapiña, M., Jimenez-Relinque, E., Castellote, M., 2014. Turning waste into valuable  
808 resource: potential of electric arc furnace dust as photocatalytic material. *Environmental*  
809 *Science and Pollution Research* 21(20), 12091-12098.  
810  
811 Sayadi, M., Hesami, S., 2017. Performance evaluation of using electric arc furnace dust  
812 in asphalt binder. *Journal of Cleaner Production* 143, 1260-1267.  
813  
814 Sebag, M.G., Korzenowski, C., Bernardes, A.M., Vilela, A.C., 2009. Evaluation of  
815 environmental compatibility of EAFD using different leaching standards. *Journal of*  
816 *Hazardous Materials* 166(2-3), 670-675.  
817  
818 Silva, P., de Brito, J., Costa, J., 2011. Viability of two new mixture design methodologies  
819 for self-consolidating concrete. *Aci Materials Journal* 108(6), 579-588.  
820  
821 Silva, R.V., de Brito, J., Evangelista, L., Dhir, R.K., 2016. Design of reinforced recycled  
822 aggregate concrete elements in conformity with Eurocode 2. *Construction and Building*  
823 *Materials* 105, 144-156.  
824  
825 Skalny, J., Johansen, V., Thaulow, N., Palomo, A., 1996. DEF: As a form of sulfate  
826 attack. *Materiales de construccion* 46(244), 5-29.  
827  
828 Tashiro, C., Oba, J., 1979. Effects of Cr<sub>2</sub>O<sub>3</sub>, Cu(OH)<sub>2</sub>, ZnO and PbO on the compressive  
829 strength and the hydrates of the hardened C<sub>3</sub>A paste. *Cement and Concrete Research* 9(2),  
830 253-258.  
831  
832 Tennich, M., Kallel, A., Ben Ouezdou, M., 2015. Incorporation of fillers from marble and  
833 tile wastes in the composition of self-compacting concretes. *Construction and Building*  
*Materials* 91(Supplement C), 65-70.

834  
835 USEPA, 2018. United States Environmental Protection Agency. Available from:  
836 <https://www.epa.gov/> (accessed September, 2018).  
837  
838 van der Sloot, H., Dijkstra, J., 2004. Development of horizontally Standardized leaching  
839 tests for construction materials: A material based or release based approach? Identical  
840 leaching mechanisms for different materials., ECN-C--04-060 ed., Available from:  
841 <http://www.ecn.nl/docs/library/report/2004/c04060.pdf> (accessed September, 2018).  
842  
843 Yousuf, M., Mollah, A., Vempati, R.K., Lin, T.C., Cocke, D.L., 1995. The interfacial  
844 chemistry of solidification/stabilization of metals in cement and pozzolanic material  
845 systems. Waste Management 15(2), 137-148.